

High Porosity in Nanostructured *n*-Type Bi₂Te₃ Obtaining Ultralow Lattice Thermal Conductivity

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KEYWORDS: *thermoelectrics, bismuth telluride, porous, nanostructure, lattice thermal conductivity*

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21 **ABSTRACT:** Porous structure possesses full potentials to develop high-performance
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24 thermoelectric materials with low lattice thermal conductivity. In this study, *n*-type porous
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27 nanostructured Bi₂Te₃ pellet is fabricated by sintering Bi₂Te₃ nanoplates synthesized with
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30 a facile solvothermal method. With adequate sublimations of Bi₂TeO₅ during the spark
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33 plasma sintering, homogeneously distributed pores and dense grain boundaries are
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36 successfully introduced into the Bi₂Te₃ matrix, causing strong phonon scatterings. From
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39 which, an ultralow lattice thermal conductivity of < 0.1 W m⁻¹ K⁻¹ is achieved in the porous
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42 nanostructured Bi₂Te₃ pellet. With the well-maintained decent electrical performance, a
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45 power factor of 10.57 μW cm⁻¹ K⁻² at 420 K, as well as the reduced lattice thermal
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48 conductivity secured a promising *zT* value of 0.97 at 420 K, which is among the highest
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51 values reported for pure *n*-type Bi₂Te₃. This study provides the insight of realizing ultralow
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3 lattice thermal conductivity by synergistic phonon scatterings of pores and nanostructure
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7 in the *n*-type Bi₂Te₃-based thermoelectric materials.
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49 INTRODUCTION

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52 Thermoelectric materials, enabling the direct solid-state conversion between heat and electricity,
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54 are promising to realize emission-free power generation and refrigeration, and alleviate current
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3 energy dilemma.¹⁻³ To evaluate efficiency of thermoelectric materials, the dimensionless figure of
4 merit (zT) is defined as: $zT = S^2\sigma T/\kappa = S^2\sigma T/(\kappa_e + \kappa_l)$, where S , σ , κ , κ_e , κ_l , and T are the Seebeck
5 coefficient, electrical conductivity, total thermal conductivity, electrical thermal conductivity,
6 lattice thermal conductivity, and the absolute temperature, respectively.⁴⁻⁶ To achieve high zT ,
7 efforts have been devoted to improving the power factor ($S^2\sigma$) through doping engineering⁷⁻⁹ or
8 band structure engineering,¹⁰⁻¹² and reducing the κ_l through nanostructure engineering.¹³⁻¹⁵

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10 Low-temperature thermoelectric materials are well-known for their high conversion efficiency
11 around the room temperature,^{16,17} and have especial attractions as the power source for wearable
12 electronics.⁵ Bismuth telluride (Bi_2Te_3), as one of the best low-temperature thermoelectric
13 materials, has intrinsically decent electrical properties due to its narrow band gap (~ 0.15 eV) with
14 high valley degeneracy and anisotropic effective mass.¹⁸ By means of alloying/doping with
15 elements,^{19,20} its electrical performance can be further boosted, and a record-high $S^2\sigma$ of $\sim 45 \mu\text{W}$
16 $\text{cm}^{-1} \text{K}^{-2}$ was reported at 300 K from both S-doped Bi_2Te_3 ²⁰ and $\text{Bi}_{0.5}\text{Sb}_{1.5}\text{Te}_3$.¹⁷ In order to further
17 enhance its zT , various efforts have been devoted to reducing κ_l by inducing strong phonon
18 scatterings.^{16,17,21} Specifically, defect engineering, including the introduction of lattice defects,
19 such as point defects¹⁶ or dislocations,¹⁷ was reported in the Bi_2Te_3 -based alloys. With the effective
20 phonon scatterings due to defect engineering, reduced κ_l of $\sim 0.3 \text{ W m}^{-1} \text{K}^{-1}$ was achieved at 300 K
21 in the $\text{Bi}_{0.5}\text{Sb}_{1.5}\text{Te}_3$.¹⁷ Moreover, nanostructure engineering was reported to induce a high density
22 of grain boundaries in the Bi_2Te_3 . As a result, significant phonon scatterings were found, leading
23 to ultralow κ_l of $\sim 0.2 \text{ W m}^{-1} \text{K}^{-1}$ at 370 K.²¹ In addition, porous structure is also known to effectively
24 reduce κ_l .²²⁻²⁴ Due to the intensive phonon scatterings at the pore-induced interfaces, ultralow κ_l of
25 $0.13 \text{ W m}^{-1} \text{K}^{-1}$ was reported at 488 K in the $\text{Bi}_2\text{Te}_{2.5}\text{Se}_{0.5}$.²⁴ In spite of these progresses, lower κ_l

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3 is still desiring for Bi₂Te₃ to achieve higher zT . According to the Callaway model, κ_l has positive
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5 relevance with the phonon relaxation time (τ), as defined below:²⁵
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$$\kappa_l = \frac{1}{3} \int_0^{\omega_{max}} C_s(\omega) v_g(\omega)^2 \tau(\omega) d\omega, \quad (1)$$

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10 where C_s is the spectral heat capacity and v_g is the phonon group velocity. Based on the Matthiessen
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12 rule,²⁶ τ is negatively related to the sum effects of different phonon scattering sources. Therefore,
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14 the combination of different phonon scattering sources can contribute to smaller τ due to full-
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16 spectrum phonon scattering, which may in turn lead to lower κ_l .
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20 Here, we prepared nanostructured n -type Bi₂Te₃ with porous structure *via* a facile solvothermal
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22 method, and subsequent sintered them into pellets through the sparking plasma sintering (SPS), as
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24 schematically shown in **Figure 1**. During the sintering process, the impurity Bi₂TeO₅ sublimed
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26 under the heating environment and large plasma currents.²⁷ Consequently, nanostructured Bi₂Te₃
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28 with porous structure was fabricated. With the synergistic effects of both pores and nanostructure,
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30 strong phonon scatterings were induced, leading to ultralow $\kappa_l < 0.1 \text{ W m}^{-1} \text{ K}^{-1}$ at 420 K.
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32 Consequently, a promising zT of 0.97 was achieved at 420 K, which is among the highest values
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34 reported for pure n -type Bi₂Te₃.^{21,22,28,29}
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42 EXPERIMENTAL SECTION

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48 Bi₂Te₃ nanoplates were synthesized *via* a typical solvothermal route. Analytical grade bismuth
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50 oxide (Bi₂O₃, 99.9%), tellurium dioxide (TeO₂, 99.999%), ethylene glycol (EG),
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52 polyvinylpyrrolidone (PVP, average molecular weight = 40000) and sodium hydroxide (NaOH,
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54 99.99%) were purchased from Sigma-Aldrich as precursors without any further purification.
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3 During the synthesis, PVP was firstly dissolved into EG (36 ml) as the surfactant, followed by 20
4 min vigorous stirring in order to form a clear solution. Bi₂O₃ and TeO₂ were then added into the
5 solution as the source materials of Bi and Te, respectively. 5 mol L⁻¹ NaOH (4 ml) was dropped
6 into the solution to provide the alkaline environment. The solution was stirred for 30 min to
7 uniformly mix precursors, before sealed in a 125 ml polytetrafluoroethylene-lined stainless-steel
8 autoclave. The autoclave was heated to 210 °C for 24 h and naturally cooled down to the room
9 temperature. Synthesized products were collected by centrifugation and washed with deionized
10 water and ethanol for several times, and finally dried in the oven at 60 °C for 12 h.
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22 **Spark Plasma Sintering and Thermoelectric Properties Measurements**

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25 In order to measure the thermoelectric properties, synthesized powders were sintered into a disc-
26 shaped pellet ($\Phi = 12.6$ mm) by SPS (SPS-211Lx, Fuji Electronic Co., Ltd.) at 300 °C for 5 min
27 under 60 MPa pressure. Density (ρ) of the pellet (85%) was measured using the Archimedes
28 method. Due to the intrinsic anisotropy of Bi₂Te₃,³⁰⁻³² thermoelectric properties of the pellets were
29 measured in the in-plane direction, since it is the high-performance direction.³² Among which, σ
30 and S were measured simultaneously using ZEM3 (ULVAC Technologies, Inc.) within the
31 temperature range between 300 and 573 K; the thermal diffusivity D was measured by LFA 457
32 (NETZSCH Group), which was subsequently applied to calculate $\kappa = D \cdot C_p \cdot \rho$,³³ where C_p is the
33 specific heat capacity measured by DSC 404 C (NETZSCH Group); n and μ were measured by a
34 Hall system under a reversible magnetic field of 1.5 T. All property measurements of the pellet
35 were conducted for at least three times in order to ensure the repeatability.
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51 **Characterization**

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Both synthesized powder and sintered pellet were systematically characterized. X-ray diffraction (XRD, Bruker-D8) was applied to determine the composition and crystal structure of both as-synthesized Bi_2Te_3 powder and as-sintered Bi_2Te_3 pellet, where the pellet was characterized along the in-plane direction. X-ray spectroscopy (XPS) (Kratos Axis Ultra) was used to investigate the valence states of Te in Bi_2Te_3 and Bi_2TeO_5 . Scanning Electron Microscopy (SEM, JEOL JSM-7100F) was utilized to observe the morphology and structural characteristics. Electron probe micro-analyzer (EPMA, JEOL JXA-8200) was used to determine the chemical composition of the pellet. The instrumental deviation of EPMA is 0.1% and ten randomly selected areas of the pellet were measured. Energy-dispersive X-ray spectroscopy (EDS) mapping and spot analysis (equipped in HITACHI-SU3500 SEM) were conducted to verify the element distribution and the chemical composition of the sample, respectively.

RESULTS AND DISCUSSION

Figure 2a shows the XRD patterns of as-synthesized Bi_2Te_3 powder and as-sintered Bi_2Te_3 pellet along the in-plane direction, respectively. Compared with standard Bi_2Te_3 diffraction pattern (PDF#15-0863), (0 0 15) diffraction peaks of both Bi_2Te_3 powder and pellet samples are found to shift towards higher degree, indicating the lattice shrinkage, as shown in Figure S1 (ESI†). The diffraction peaks of the powder can be exclusively indexed as the rhombohedral structured Bi_2Te_3 with the lattice parameters of $a = 4.3852$ nm, $c = 30.483$ nm and a space group of $R\bar{3}m$ (PDF#15-0863), and orthorhombic structured Bi_2TeO_5 with the lattice parameters of $a = 16.447$ nm, $b = 5.513$ nm, $c = 11.579$ nm and a space group of $\text{Cm}2a$ (PDF#38-0420). However, after the SPS process, diffraction peaks of as-sintered pellet can be only indexed as the rhombohedral structured

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3 Bi_2Te_3 . The disappearance of Bi_2TeO_5 during the sintering process should be attributed to the
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5 sublimation of Bi_2TeO_5 under SPS heat treatments and large currents.²⁷ The XPS was further
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7 utilized to verify the sublimation of Bi_2TeO_5 during the SPS process. The full XPS spectra of both
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9 Bi_2Te_3 nano-powder and nanostructured Bi_2Te_3 pellet (shown in Figure S2, ESI†) indicate the
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11 presences of Bi 4f and Te 3d energy states, without any energy states detected for other elements,
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13 except C 1s and O 1s. High-resolution scans of XPS spectra for Bi 4f show no obvious difference
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15 after the sintering, where peaks of Bi 4f_{5/2} and Bi 4f_{7/2} were found at 162.5 eV and 157.2 eV,
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17 respectively, as shown in Figure S2, ESI†. This implies the existence of single valence state Bi^{3+}
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19 in both Bi_2Te_3 nano-powder and porous nanostructured Bi_2Te_3 pellet.³⁷ **Figure 2b** and **c** show the
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21 high-resolution scans of XPS spectra for Te, in which strong singlet peaks of Te 3d_{5/2} and Te 3d_{3/2}
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23 were detected at the binding energy of 572.23 eV and 582.53 eV in both powder and pellet,
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25 respectively. This indicates the existence of Te^{2-} .^{34,35} Additional peaks at the binding energy of
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27 575.67 eV and 586.14 eV, evidencing the existence of Te^{4+} , were only found in the powders.³⁵ The
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29 Te^{2-} and Te^{4+} should be ascribed to Bi_2Te_3 and Bi_2TeO_5 , respectively. The disappearance of Te^{4+}
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31 peaks after the SPS process further confirms the sublimation of Bi_2TeO_5 .
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37 **Figure 3a-c** show SEM images of as-synthesized Bi_2Te_3 powder. As can be seen, as-synthesized
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39 Bi_2Te_3 powders are consisted of hexagonal nanoplates, with a typical thickness of ~20 nm, as
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41 shown in the inset of Figure 3c. Such nanostructures can induce a high density of grain boundaries
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43 in the pellet after the sintering.²⁹ Figure 3d-3f show SEM images of as-sintered nanostructured
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45 Bi_2Te_3 pellet. As highlighted in the image, porous structure can be clearly seen, where pores
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47 homogeneously distribute in the matrix. Figure 3f shows a typical pore with a size of ~400 nm in
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3 **Figure 4a** show the EDS results obtained from as-sintered nanostructured porous Bi_2Te_3 pellet.
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5 EDS spot analysis (spot highlighted in the inset) suggests that the sample show binary composition,
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7 with Bi and Te exclusively detected. EDS mapping and corresponding backscattered electrons
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9 (BSE) SEM image are shown as the inset, where homogeneous distributions of Bi and Te were
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11 observed. The atomic ratios of Bi and Te were evaluated as 41.2 % and 58.8 %, respectively.
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14 Figure 4b shows the EPMA results, where the composition of as-sintered nanostructured porous
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16 pellet was more accurately determined. The average ratio between Bi and Te in the pellet is ~ 0.715 ,
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18 indicating a composition of $\text{Bi}_2\text{Te}_{2.8}$. Such a Te vacancy is consistent with the observed lattice
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20 shrinkage in our XRD results. Detailed compositions of Bi and Te detected by EPMA can be found
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22 from Table S1 (ESI[†]) in the supporting information.
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26 **Figure 5** shows the measured thermoelectric properties of as-sintered nanostructured porous
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28 Bi_2Te_3 pellet along the in-plane direction, including σ , n , μ , S and κ . In order to understand the
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30 effects of porous structure, thermoelectric properties are compared with previously reported dense
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32 nanostructured Bi_2Te_3 pellet.²¹ As shown in Figure 5a, σ shows no obvious difference than that of
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34 the dense nanostructured Bi_2Te_3 pellet, and decreased gradually with T , presenting typical metallic
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36 transport behaviours.³⁶ The peak σ of 770 S cm^{-1} , occurred at 300 K, is comparable to other
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38 reported pure Bi_2Te_3 .^{21,28,37} Figure 5b shows measured n and μ , in which n slightly increases from
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40 $7 \times 10^{19} \text{ cm}^{-3}$ to $1 \times 10^{20} \text{ cm}^{-3}$ in the temperature range of 300 to 550 K, while μ gradually decreases
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42 with T , with the value of $66 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ at 300 K. Such a μ is similar to other porous Bi_2Te_3 -based
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44 pellets.^{24,38} In terms of S , as shown in Figure 5c, negative values were obtained, indicating typical
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46 n -type properties. The peak value of $-145 \text{ } \mu\text{V K}^{-1}$ was measured at $\sim 450 \text{ K}$, which is also similar
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48 to that of the dense nanostructured Bi_2Te_3 .^{21,37} As a result, Figure 5d shows uncompromised $S^2\sigma$
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50 that has a peak value of $12.89 \text{ } \mu\text{W cm}^{-1} \text{ K}^{-2}$ at 320 K. It can be concluded that the introduction of
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3 pores possesses negligible effects on electrical properties of our nanostructured porous Bi_2Te_3
4 pellet along the in-plane direction. This may be ascribed to the typical layered structure of Bi_2Te_3 ,
5 where the carrier transport in the in-plane direction (ab direction) primarily contributes to the
6 system carrier transport, compared with that in the out-of plane direction (c direction).³⁰⁻³² The
7 pores observed here have limited influences on the atomic scale layered structure. Hence, the high
8 carrier transport properties can be secured by nearly undisturbed layered structures.

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10 **Figure 6** shows calculated κ . The results of measured D and C_p from 300 K to 550 K can be found
11 from Figure S3, ESI†. It was found that the introduction of pores leads to lower κ at all temperature
12 range from 300 to 560 K, with the lowest value of $\sim 0.5 \text{ W m}^{-1} \text{ K}^{-1}$ at 418 K. To understand such
13 an ultralow κ , we investigated κ_e and κ_l using the Wiedemann – Franz law,³⁹ where $\kappa_e = L\sigma T$ and
14 $\kappa_l = \kappa - \kappa_e$. L is the Lorenz number ($\sim 1.8 \times 10^{-8} \text{ v}^2 \text{ K}^{-2}$) and was calculated based on SPB model
15 (Figure S3, ESI†).^{40,41} It was found that κ_e lies in the range of 0.39 to 0.47 $\text{W m}^{-1} \text{ K}^{-1}$ (Figure S3,
16 ESI†), which is comparable to the values of dense nanostructured Bi_2Te_3 .²¹ Therefore, the
17 reduction of κ of the as-sintered nanostructured porous Bi_2Te_3 pellet mainly derives from κ_l . Figure
18 6a shows calculated κ_l in as-sintered porous nanostructured Bi_2Te_3 pellet. It was found porous
19 structure can effectively reduce κ_l and ultralow κ_l of $< 0.1 \text{ W m}^{-1} \text{ K}^{-1}$ was found at 370 K. This
20 significant κ_l reduction can be explained by the phonon gas theory, where κ_l can be written as $\kappa_l =$
21 $(1/3)C_v v l$. Among which, C_v is the specific heat capacity at a constant volume, v is the sound
22 velocity, and l is the phonon mean free path.⁴² Reduced ρ of the porous pellet can negatively
23 contribute to C_v , considering $C_v = C'_v \rho$, where C'_v reflects the specific heat properties of the
24 materials.⁴³ Moreover, formation of pores has been proved to reduce v and l of thermoelectric
25 materials.⁴³ As a result, κ_l was significantly reduced in our porous structure, and in turn lead to
26 ultralow κ .⁴⁴⁻⁵¹ Benefited from uncompromised $S^2\sigma$ and effectively reduced κ , we ultimately

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3 realized boosted zT near 1 at 370 K, as shown in the Figure 6b. Such a promising zT is among the
4 highest values reported for pure n -type Bi_2Te_3 so far, as shown in Figure S4 (ESI†).^{21,28,29} Figure
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6 6c compared our results with previously reported porous²² or nanostructured Bi_2Te_3 ,²¹ it was found
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8 that the simultaneous introduction of the nanostructure and the porous structure in our study can
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10 effectively generate further reduced κ_l , which in turn produce improved zT . Based on the analysis
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12 of single parabolic band (SPB) model as shown in Figure 6d,^{40,41} further reduced κ_l could
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14 potentially pushes zT to a higher level with appropriately engineered n_e (Section 5 in the Supporting
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16 Information shows detailed calculations). The higher n_e than predicted optimal value in our work
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18 may be associated with the high level of Te vacancy. Further control of Te vacancy might be of
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20 interest in our future study to effectively reduce n_e and in turn achieve higher zT .
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30 CONCLUSIONS

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33 In summary, we fabricated n -type porous nanostructured Bi_2Te_3 pellet with facile solvothermal
34 method and subsequent SPS. With adequate sublimations of Bi_2TeO_5 during the sintering process,
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36 homogeneously distributed pores were successfully introduced into the nanostructured Bi_2Te_3
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38 pellet. As a consequence of simultaneous strong phonon scatterings at both dense grain boundaries
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40 of nanostructure and interfaces induced by pores, an ultralow κ_l ($< 0.1 \text{ W m}^{-1} \text{ K}^{-1}$) was realized.
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42 The well-maintained decent electrical performance, a $S^2\sigma$ of $10.57 \mu\text{W cm}^{-1} \text{ K}^{-2}$ at 420 K, as well
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44 as the reduced κ_l secured a promising zT of 0.97, which is among the highest values reported for
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46 pure n -type Bi_2Te_3 .^{21,22,28,29} This study provides an alternative insight of realizing ultralow κ_l by
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48 synergistic phonon scatterings induced by pores and nanostructure in n -type Bi_2Te_3 -based
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50 thermoelectric materials.
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7 **ASSOCIATED CONTENT**
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10 **Supporting Information.** XPS spectra; Lorenz factor and measured thermal properties;
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14 EPMA results; and SPB calculation details.
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21 **AUTHOR INFORMATION**
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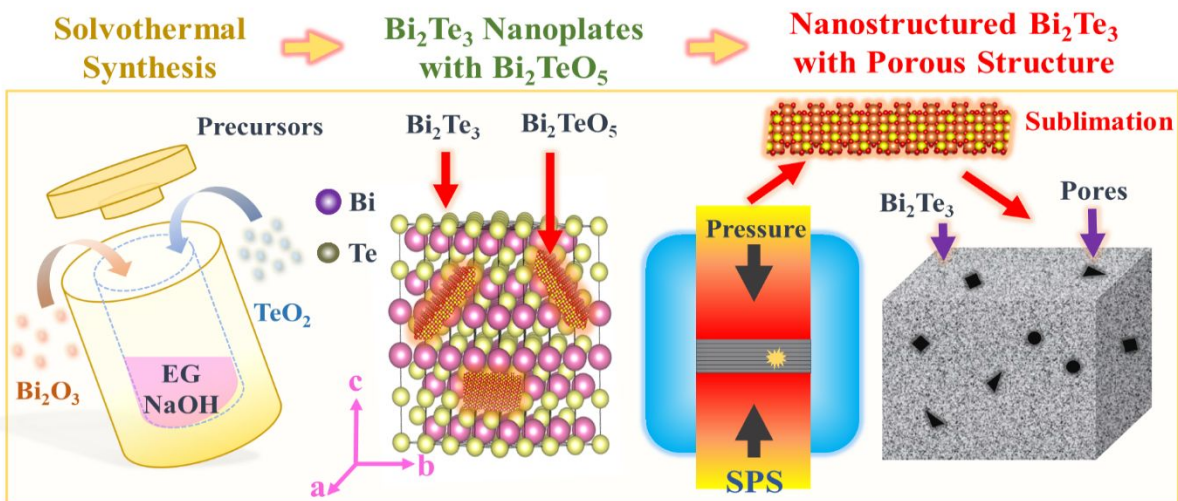


Figure 1. Illustration of the fabrication processes, where solvothermal synthesis and spark plasma sintering were applied to fabricate porous nanostructured Bi_2Te_3 pellet.

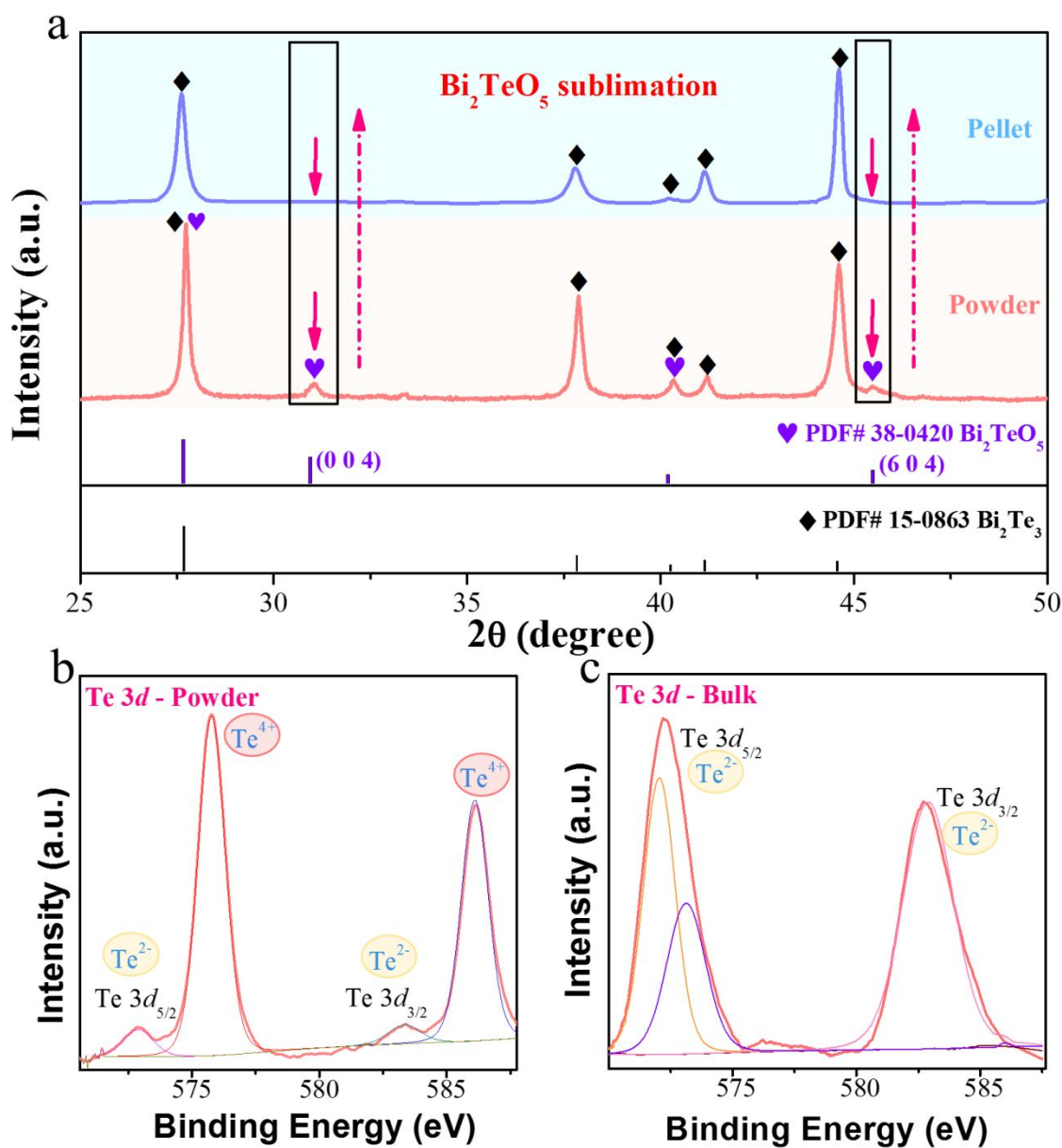


Figure 2. (a) XRD patterns of both as-synthesized Bi₂Te₃ nano-powders and as-sintered porous nanostructured Bi₂Te₃ pellet. High-resolution XPS scan of Te in the (b) as-synthesized Bi₂Te₃ nano-powder and (c) as-sintered porous nanostructured Bi₂Te₃ pellet.

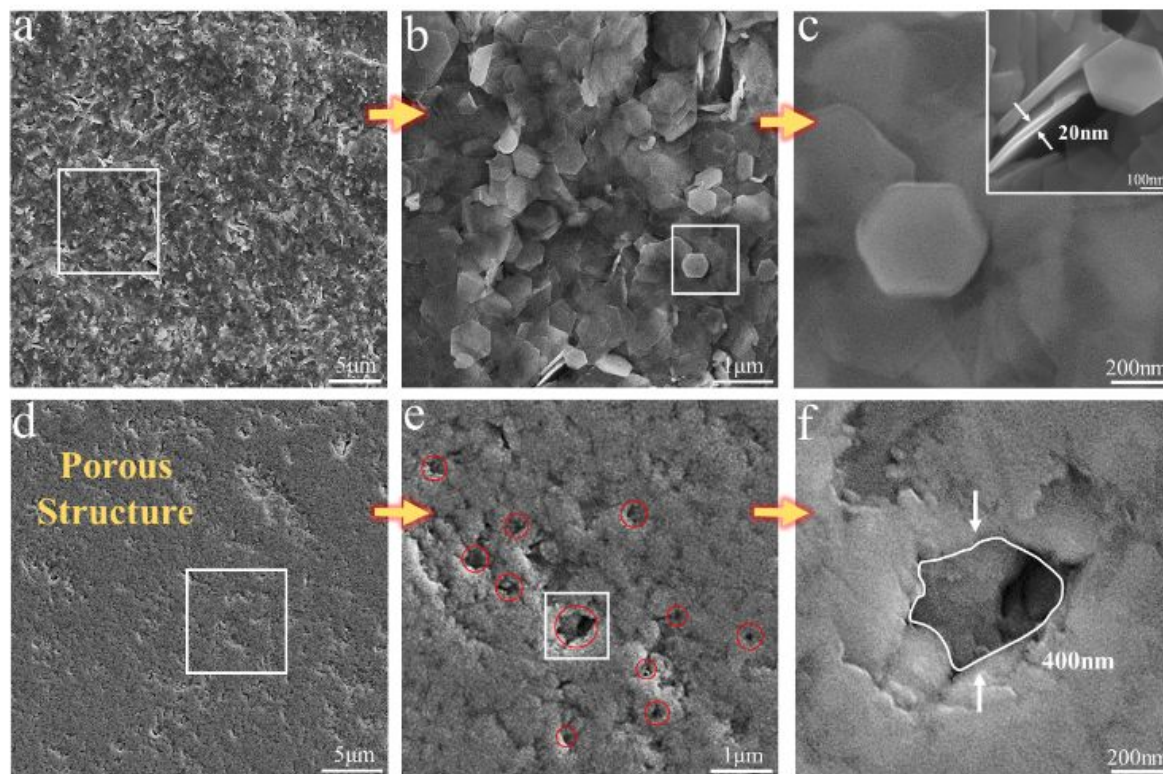
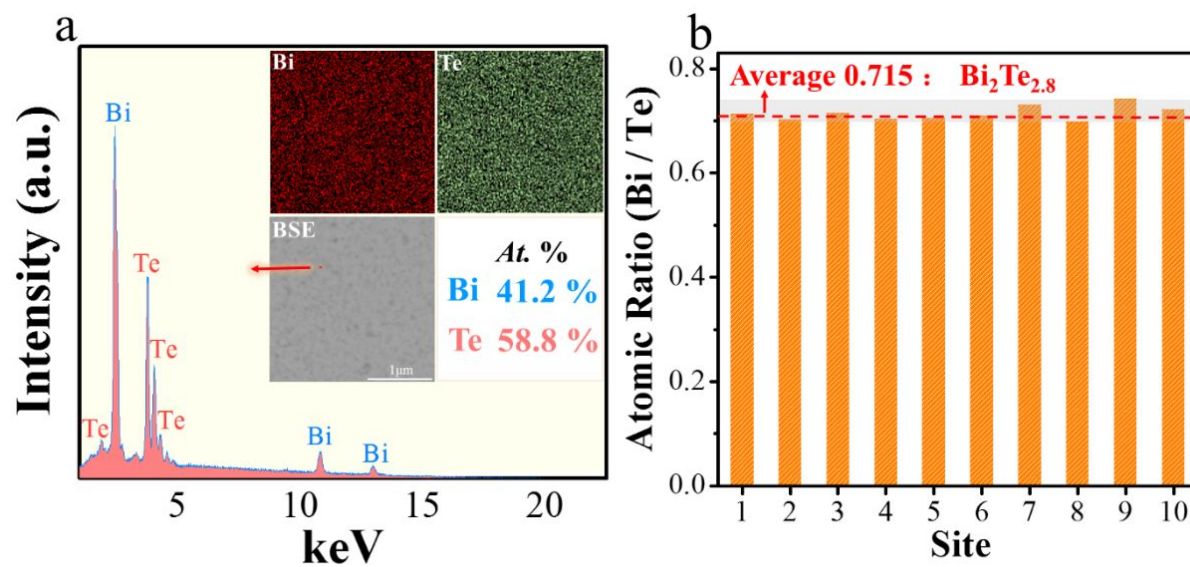


Figure 3. (a) SEM image of as-synthesized Bi_2Te_3 nano-powder; (b) magnified SEM image of the area selected in (a); and (c) magnified SEM image of the area selected in (b), showing the hexagonal plate-like nanostructure. Inset is the SEM image showing the typical lateral thickness of the nanoplates is around 20 nm. (d) SEM image of as-sintered nanostructured porous Bi_2Te_3 ; (e) magnified SEM image of the area selected in (d); and

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3 (f) magnified SEM image of the pore selected in (e), showing the typical pore size of 400
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52 **Figure 4.** (a) EDS spot spectrum of as-sintered nanostructured porous Bi₂Te₃ pellet with
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55 the inset of mapping results of Bi and Te. (b) Atomic ratios between Bi and Te measured
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4 by EPMA at ten randomly selected area in the as-sintered nanostructured porous Bi_2Te_3
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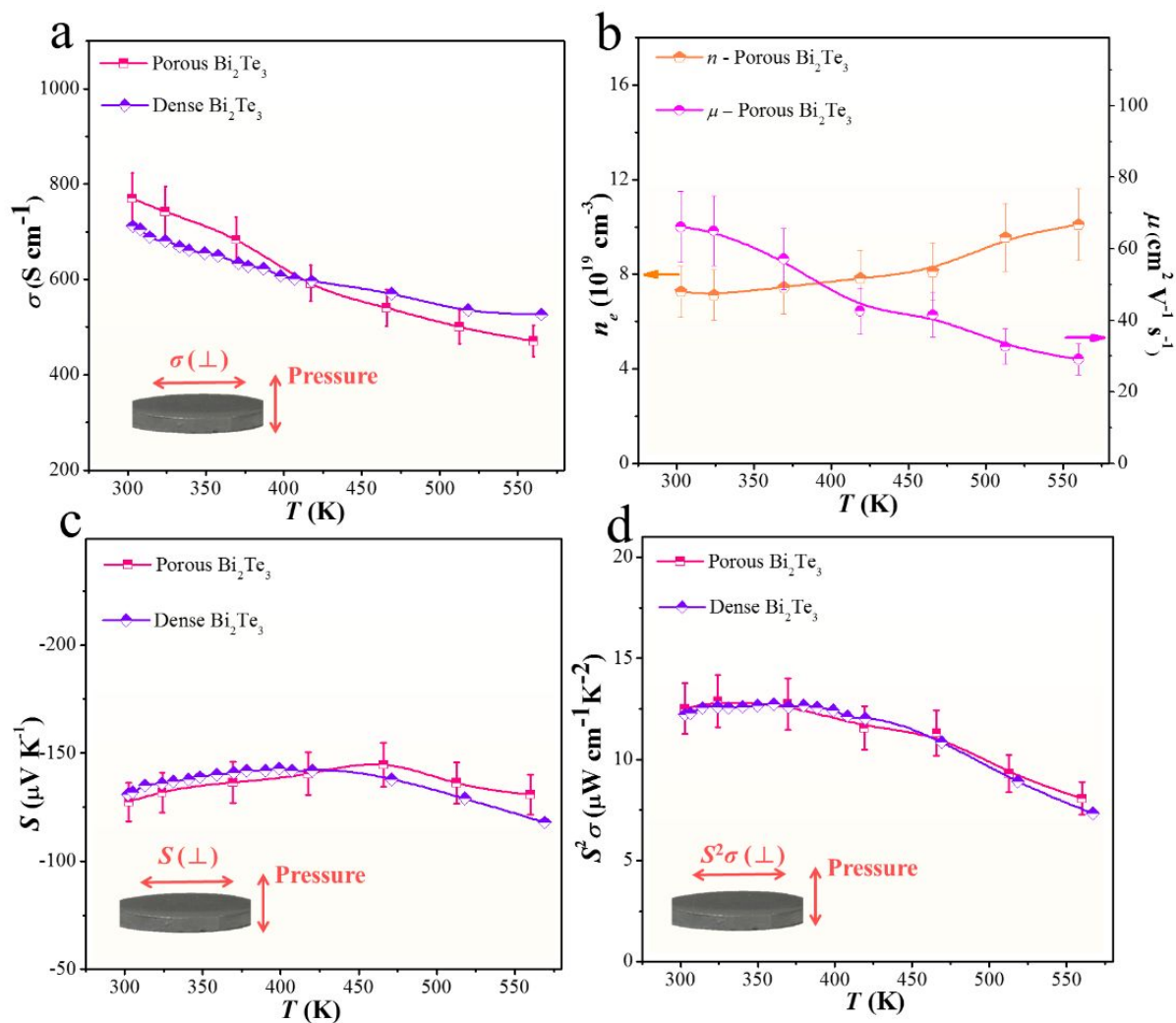


Figure 5. T -dependent (a) σ , (b) n and μ , (c) S and (d) $S^2\sigma$ of as-sintered nanostructured porous Bi_2Te_3 pellet compared with dense nanostructured Bi_2Te_3 .⁹

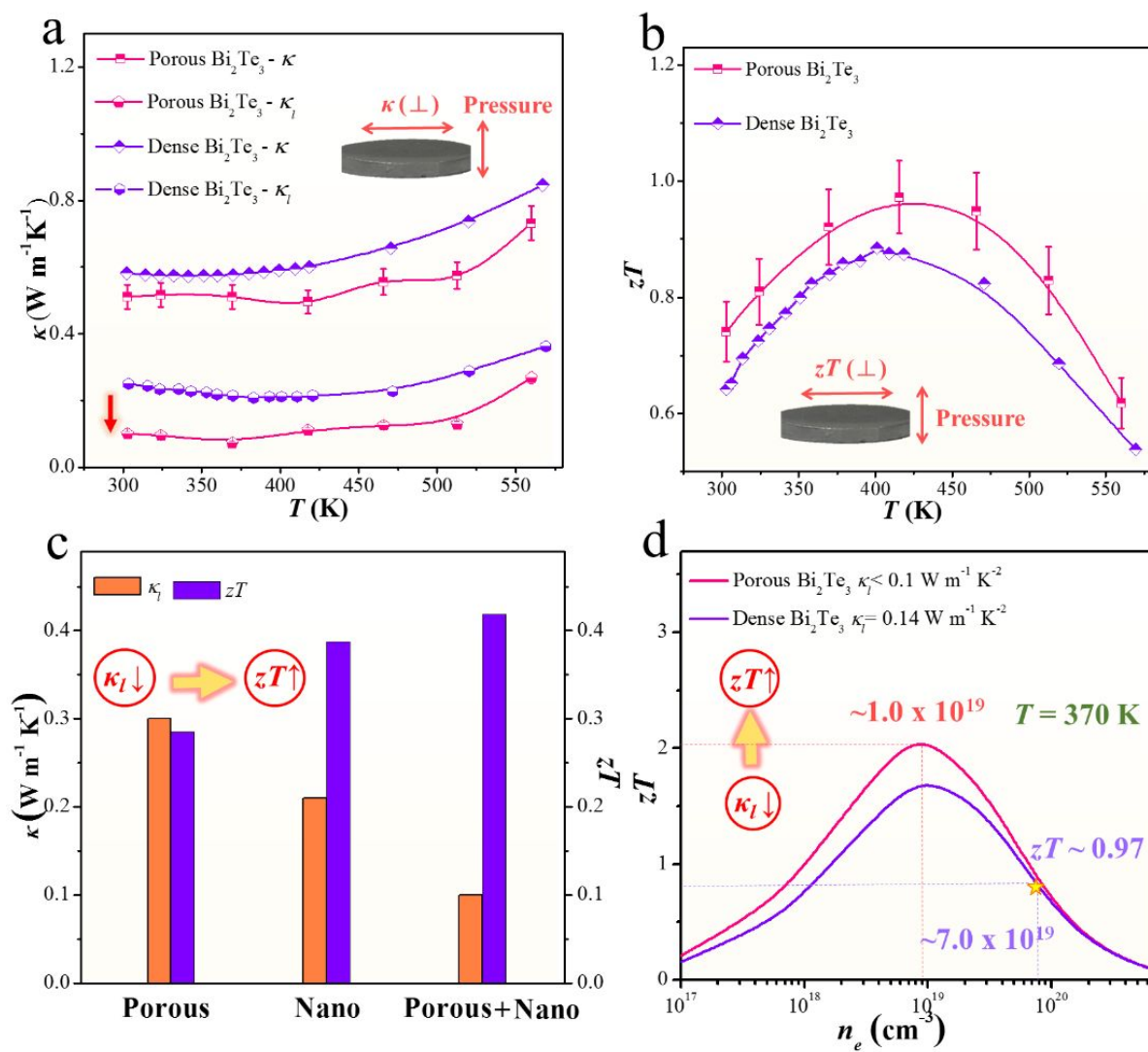


Figure 6. T -dependent (a) κ and κ_i and (b) zT of as-sintered nanostructured porous Bi₂Te₃

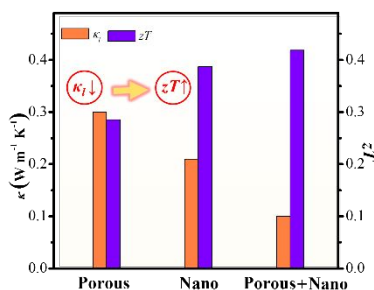
pellet compared with dense nanostructured Bi₂Te₃.⁹ (c) Comparisons of κ_i and zT

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3 between this work and previously reported porous^{10a} or nanostructured Bi₂Te₃.⁹ (d)

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7 Calculation results of SPB showing the possibility to further enhance zT with controlled

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10 n_e and reduced κ_l .

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30 **Porous nanostructured Bi₂Te₃**



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