

### USQ Combustion Meeting 26 Nov 2012

### PREMIXED COMBUSTION OF COCONUT OIL IN A HELE-SHAW CELL

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#### **OUTLINE OF PRESENTATION**

INTRODUCTION

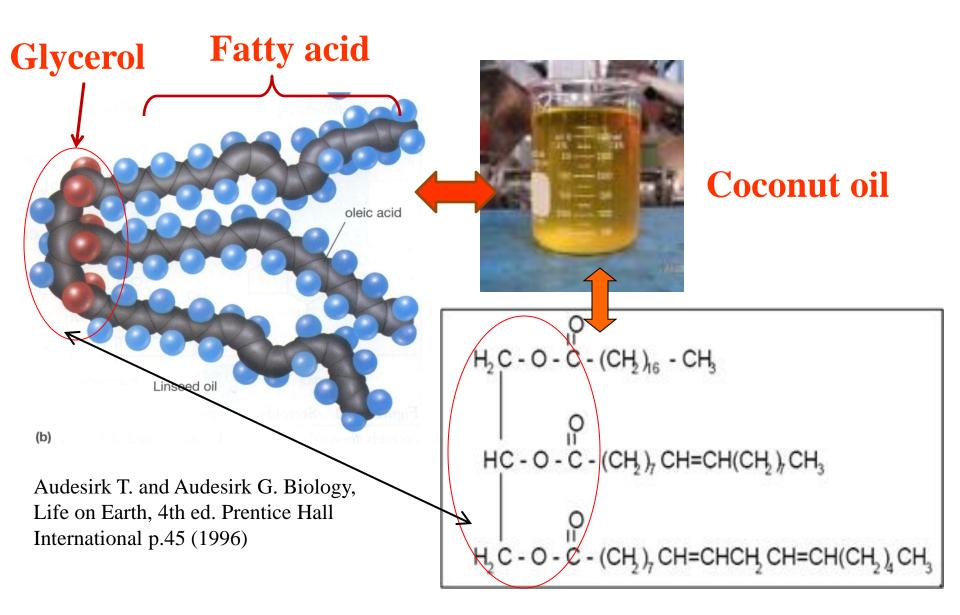
EXPERIMENTS

RESULTS AND DISCUSION

CONCLUSION

#### **MOLECULAR STRUCTURE**







#### COCONUT OIL

Energy content: 3,607 kJ (862 kcal)/100g

Viscosity: 10 x fossil diesel fuel

 Comonly be used for Diesel Engine (Stright Vegetable Oil)

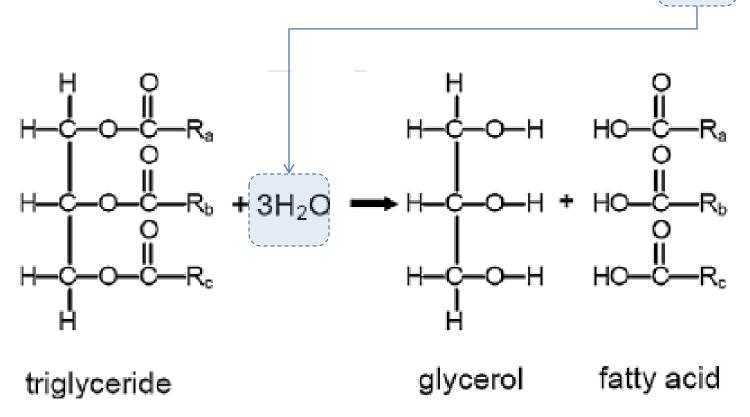


#### **FUTURE GOAL**

 To promote premixed combustion technology for coconut oil

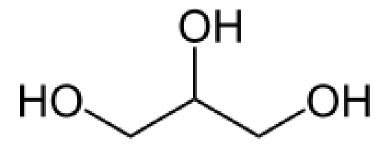
#### HYDROLYSIS REACTION OF TRIGLYCERIDE

COMBUSTION REACTION:  $aC_xH_yO_z + bO_2 \rightarrow cCO_2 + dH_2O$ 





#### **GLYCEROL**



Molecular formula: C<sub>3</sub>H<sub>8</sub>O<sub>3</sub>

<u>hygroscopic</u>

Flash point: 160 °C (closed cup)

**176** °C (open cup)

Boiling point: 290 °C

Molar mass: 92.09 g mol<sup>-1</sup>



### Main Fatty acid in coconut oil

Fatty Acid	Composition/ Molecular Formula	%	Flash point (°C)
Lauric acid	C 12:0 / C <sub>12</sub> H <sub>24</sub> O <sub>2</sub>	45.1 - 53.2	≥110
Myristic acid	C 14:0 / C <sub>14</sub> H <sub>28</sub> O <sub>2</sub>	16.8 - 21.0	_



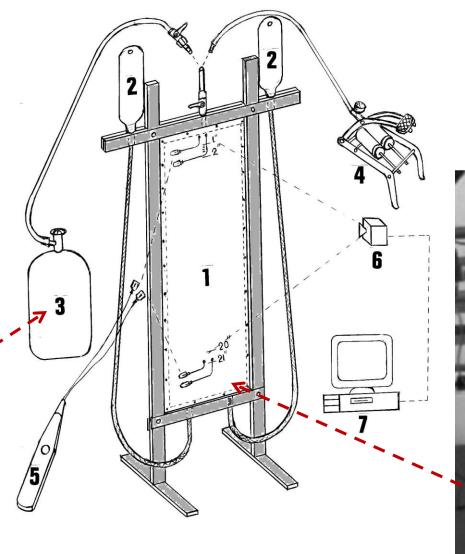
#### RESEARCH OBJECTIVE

 To elucidate Combustion Behavior of MULTI COMPONENTS FATTY ACIDS AND GLYCEROL IN COCONUT OIL

### **EXPERIMENT**



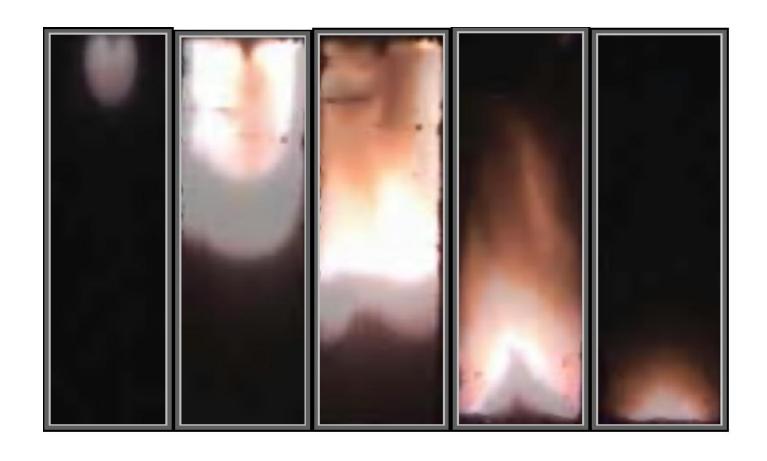
AFR= 5.3 to 7.4





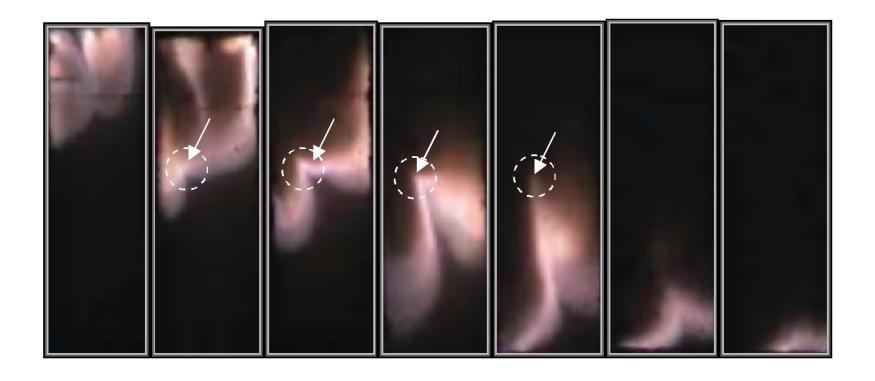
# Downward flame propagation (AFR = 5.3)



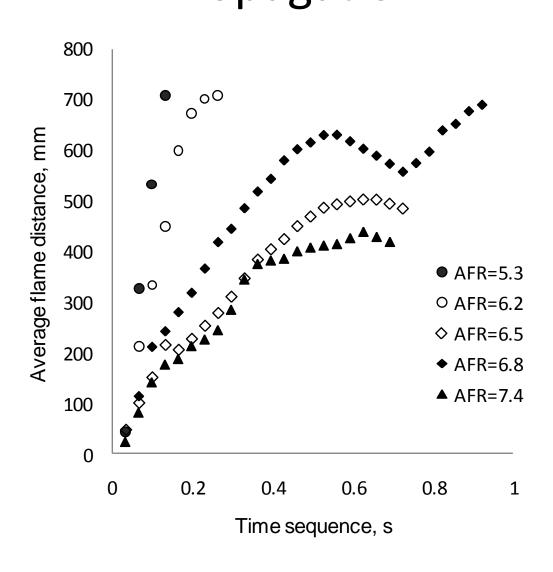




### Downward flame propagation (AFR = 6.2)

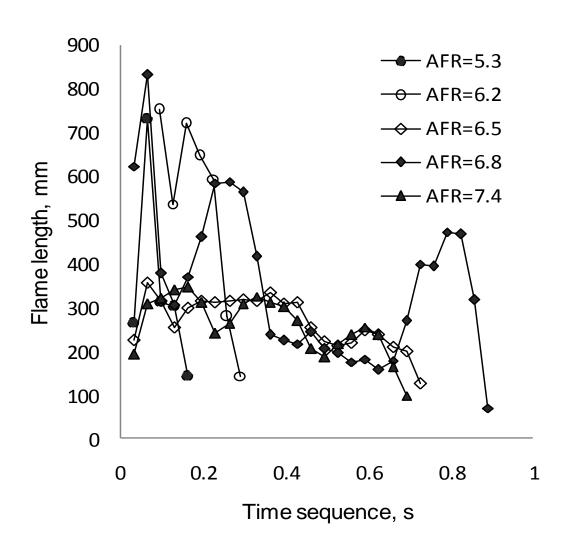


### Average Flame Distance in Downward Propagation



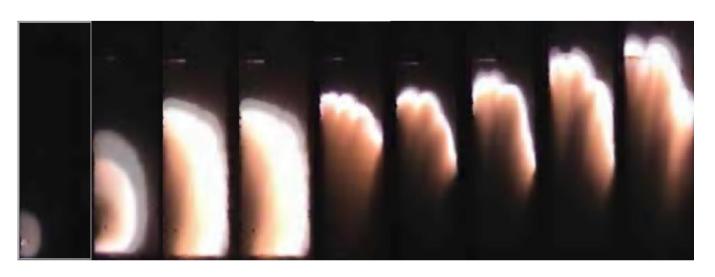
### Flame Length in Downward Propagation





# Upward Flame Propagation (AFR = 5.3)

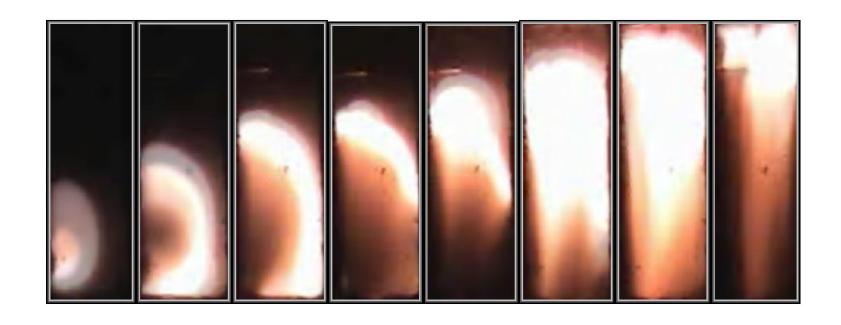






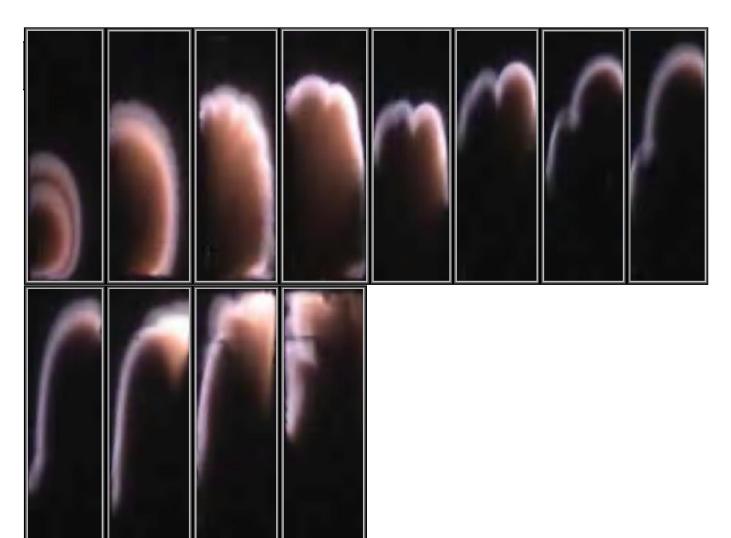
# Upward Flame Propagation (AFR = 6.2)





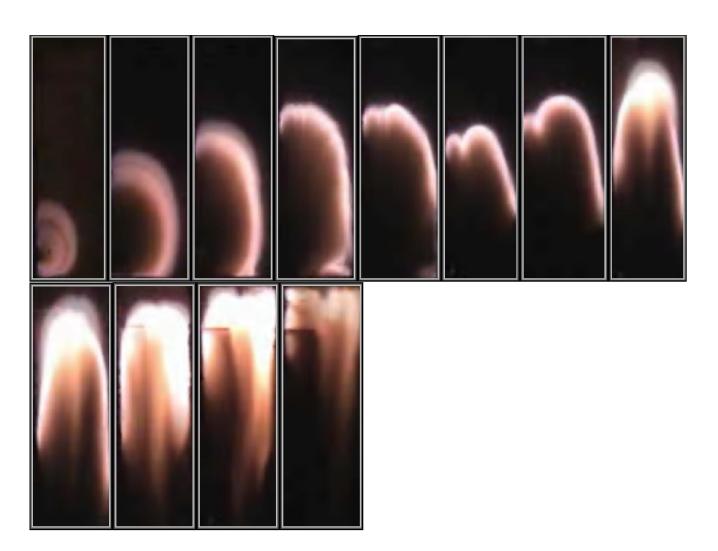
# Upward Flame Propagation (AFR = 6.8)



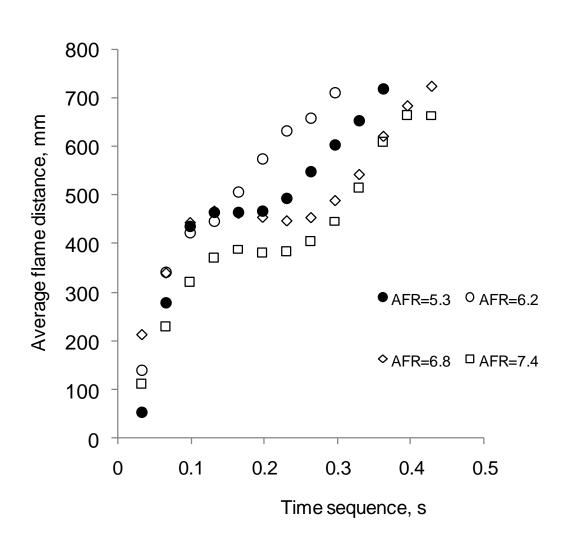


# Upward Flame Propagation (AFR = 7.4)

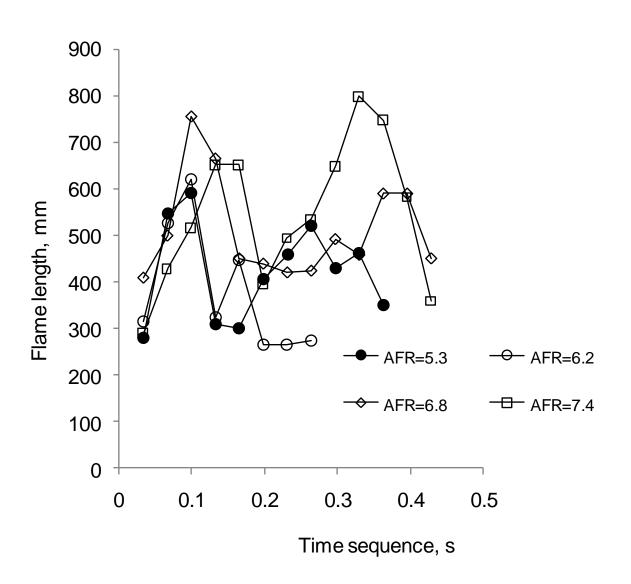




### Average Flame Distance in Upward Propagation



#### Flame Length in Upward Propagation





#### CONCLUSION

- Coconut oil tends to break into glycerol and fatty acid. Fatty acid burns first and glycerol does later.
- Fatty acid combustion produced separated
   flame which is clearly visible at lean mixture
- The ignition of glycerol needs substatial heating that slows down flame propagation and tends to EXPLODE